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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/538,477	06/07/2005	Chih-Chang Chu	CHUC3007	2280
23364 BACON & THO	7590 11/24/200 OMAS, PLLC	EXAMINER		
625 SLATERS	LANE	HAIDER, SAIRA BANO		
FOURTH FLOOR ALEXANDRIA, VA 22314-1176			ART UNIT	PAPER NUMBER
			1796	
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			11/24/2008	PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

	Application No.	Applicant(s)			
	10/538,477	CHU ET AL.			
Office Action Summary	Examiner	Art Unit			
	SAIRA HAIDER	1796			
The MAILING DATE of this communication app Period for Reply	ears on the cover sheet with the c	orrespondence address			
A SHORTENED STATUTORY PERIOD FOR REPLY WHICHEVER IS LONGER, FROM THE MAILING DA - Extensions of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication. - If NO period for reply is specified above, the maximum statutory period w - Failure to reply within the set or extended period for reply will, by statute, Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	ATE OF THIS COMMUNICATION 36(a). In no event, however, may a reply be tim vill apply and will expire SIX (6) MONTHS from cause the application to become ABANDONE	N. nely filed the mailing date of this communication. D (35 U.S.C. § 133).			
Status					
Responsive to communication(s) filed on 14 Ju This action is FINAL . 2b)☑ This Since this application is in condition for allowar closed in accordance with the practice under E	action is non-final. nce except for formal matters, pro				
Disposition of Claims					
4) Claim(s) 1-7 is/are pending in the application. 4a) Of the above claim(s) 8 is/are withdrawn fro 5) Claim(s) is/are allowed. 6) Claim(s) 1-7 is/are rejected. 7) Claim(s) is/are objected to. 8) Claim(s) are subject to restriction and/or Application Papers 9) The specification is objected to by the Examine 10) The drawing(s) filed on is/are: a) access applicant may not request that any objection to the or	relection requirement. r. epted or b)□ objected to by the B				
Replacement drawing sheet(s) including the correcti 11) The oath or declaration is objected to by the Ex-		• •			
Priority under 35 U.S.C. § 119					
12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) All b) Some * c) None of: 1. Certified copies of the priority documents have been received. 2. Certified copies of the priority documents have been received in Application No 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received.					
Attachment(s) 1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date 6/10/2008.	4) Interview Summary Paper No(s)/Mail Da 5) Notice of Informal P 6) Other:	nte			

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DETAILED ACTION

Election/Restrictions

1. Applicant's election with traverse of Group I: Claim 1-7 in the reply filed on 7/14/2008 is acknowledged. The traversal is on the ground(s) that claim 8, as amended, should be treated as a member of Group I. This is not found persuasive because claim 8, as amended, which is drawn to the microspheres product, does not require the limitation of the method recited in the claims of Group I.

The requirement is still deemed proper and is therefore made FINAL.

Claim Rejections - 35 USC § 103

- 2. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 3. The factual inquiries set forth in *Graham* v. *John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:
 - 1. Determining the scope and contents of the prior art.
 - 2. Ascertaining the differences between the prior art and the claims at issue.
 - 3. Resolving the level of ordinary skill in the pertinent art.
 - 4. Considering objective evidence present in the application indicating obviousness or nonobviousness.
- 4. Claim 1 is rejected under 35 U.S.C. 103(a) as being unpatentable over Ekman et al. (US 4,822,535) in view of Moiser (US 4,492,720).

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- 5. Ekman discloses method of producing small spherical polymer particles from systems containing two liquid phases, the one phase of which contains one or more dissolved substances and is dispersed in the form of small droplets in the other phase to form an emulsion, whereafter the droplets are converted to a solid form. The liquid phases used are two mutually immiscible aqueous phases (abstract). Ekman notes that whole (living) cells, cell organelles, solid particles or small oil droplets can be encapsulated when practicing the invention (col. 8, lines 15-17).
- 6. In example 6, Ekman discloses the preparation of spherical particles of cross-liked dextran. The process involves preparing a first aqueous solution of acryldextran (M_w =40,000). Ekman discloses that acryldextran which functions as both the monomer and a crosslinker. Next, a second aqueous solution is then prepared from polyethylene glycol (M_w =6,000). The first aqueous solution is added to the second aqueous solution and emulsified, wherein the first aqueous solution is the inner phase (the droplets). Polymerization is initiated via a catalyst. The particles are collected via filtration.
- 7. Ekman fails to explicitly disclose the size of the droplets; however Ekman notes that the particle size of the solid particles obtained can be controlled in all of the disclosed embodiments in a manner known per se, for example by stirring with varying intensities or by selecting suitable viscosities for the various phases. In the case of the system polyethylene glycol-starch the particle size can also be regulated by selection of the molecular weight of the polyethylene glycol, a polyethylene glycol of higher molecular weight providing larger particles (col. 7, lines 66 to col. 8, lines 4). Thus attention is directed to the Moiser reference, which discloses a method of preparing microspheres for intravascular delivery. Specifically, Moiser creates the microspheres via formation of an emulsion of two phases, wherein the droplets that comprise the dispersed phase have average sizes in the range of 50-150 microns (col. 4, lines 3-26). Wherein the resulting microspheres are in

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the range of 50-350 microns and are rendered suitable for administration of therapeutic agents and diagnostic agents via intra-arterial delivery (col. 1, lines 49-59). Accordingly, it would have been obvious to one of ordinary skill in the art at the time of the invention to alter the Ekman process conditions (as taught by Ekman) in order to form droplets and resulting microcapsules suitable for intra-arterial delivery, as taught by Moiser.

- 8. Claim 2 is rejected under 35 U.S.C. 103(a) as being unpatentable over Ekman et al. (US 4,822,535) in view of Moiser (US 4,492,720), and in view of Jahns (US 5,596,051).
- 9. Ekman applies as above and notes that polyethylene glycol is preferably the continuous phase (col. 3, lines 19-24); however Ekman discloses that suitable two-phase systems of polymeric aqueous solutions include dextran/polyethylene glycol/water and polyethylene glycol/dextran sulphate/water (col. 1, line 66 to col. 2, line 19). Accordingly, it is readily envisaged that polyethylene glycol is the dispersed phase and dextran is the continuous phase.
- 10. It is noted that Ekman discloses polyethylene glycol as the dispersed phase, but fails to disclose the claimed polyethylene glycol diacrylate, as claimed. Thus attention is directed to the Jahns reference, which discloses the formation of microcapsules which may contain drugs and enzymes as the core material (abstract; col. 4, limes 21-22). Specifically, Jahns discloses polyethylene glycol diacrylate as suitable crosslinking monomers for the microcapsule shell (col. 3, lines 5-11). Jahns recognizes that such crosslinking monomers swells the microcapsule such that the core material can be released over a longer period of time (col. 3, lines 26-39). Further it is noted that Ekman prefers the usage of crosslinking monomers for the dispersed phase (Example 6). Accordingly, it would have been obvious to one of ordinary skill in the art at the time of the invention to utilize

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poly(ethylene glycol) diacrylate as the dispersed phased in the hydrogel formation process taught by Ekman and Moiser in order to forma sustained release microcapsule.

11.

- 12. Claims 3-7 are rejected under 35 U.S.C. 103(a) as being unpatentable over Ekman et al. (US 4,822,535) in view of Moiser (US 4,492,720), and in view of Jahns (US 5,596,051) and in further view of Nelson (US 6,596,296).
- 13. In reference to claims 3-4, Ekman applies as above but fails to disclose the claimed second hydrogel precursor as N-isopropylacrylamide. Thus attention is directed to the Nelson reference, which discloses drug releasing biodegradable fiber implants. Specifically, Nelson discloses polymer hydrogel nanospheres loaded with biological molecules, wherein useful polymer hydrogels include N-isopropylacrylamide (NIPA). Nelson recognizes NIPA gels as having the ability to undergo dramatic volume changes of 100 fold in response to small (2-3°C) temperature change; specifically, the phase transition can be adjusted to occurs at 38-39 °C such that the nanosphere is responsive to the physiological state of the patient. The nanospheres release the drug in response to an increase in the body temperature of a patient. (Example 4 at col. 20, line 40-67). Accordingly, it would have been obvious to one of ordinary skill in the art at the time of the invention to include NIPA in the polymer hydrogel taught by Ekman, Moiser, and Jahns above in order to form a composition which readily releases the active core component upon a temperature change.
- 14. In reference to claims 5-7, it is noted that Ekman discloses these limitations. Specifically, Ekman discloses dextran as a suitable continuous phase and Ekman exemplifies utilization of dextran with M_w of 40,000 (Example 4). Further, Ekman discloses the inclusion of water soluble salts such as magnesium sulphate which will reduce the solubility of dextran in the water (col. 2, lines

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21-36). Wherein utilization of such disclosure of Ekman in the invention taught by the combination

of references would be obvious to one of ordinary skill in the art.

Any inquiry concerning this communication or earlier communications from the examiner

should be directed to SAIRA HAIDER whose telephone number is (571)272-3553. The examiner

can normally be reached on Monday-Friday from 10am-6pm.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor,

Randy P. Gulakowski can be reached on (571) 272-1302. The fax phone number for the

organization where this application or proceeding is assigned is 571-273-8300.

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/Randy Gulakowski/

Supervisory Patent Examiner, Art Unit 1796

Saira Haider

Examiner

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